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$\text{C}_2\text{H}_5\text{OH}(l) + 3 \text{ O}_2(g) \rightarrow 3 \text{ H}_2\text{O}(g) + 2 \text{ CO}_2(g)$  Cot 31) A 100.0 Ml. Sample Of 0.300 M NaOH Is Mixed With A 100.0 Ml. Sample Of 0.305 M  $\text{HNO}_3$  31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 Oc And The Temperature Of The Resulting Solution Was Recorded As 37.0 Cc, Determine The  $\Delta H_{\text{rxn}}$  (in Units Of KJ/mol). Jan 20th, 2024

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