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### **Colligative Properties Colligative Properties**

3 Another Example A Solution With Two Volatile Components • Modified Raoult's Law •  $VP_T = X_A P_A^* + X_B P_B^*$  Example • At 20.0 °C The Vapor Pressures Of Methanol ( $CH_3OH$ ) and Ethanol ( $C_2H_5OH$ ) Are 95.0 And 45.0 mmHg Respectively. An Ideal Solution Contains 16.1 g Of Methanol 5th, 2024

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Excudent Alii Spirantia Mollius Aera (credo Equidem), Uiuos Ducent De Marmore Uultus, Orabunt Causas Melius, Caelique Meatus Describent Radio Et Surgentia Sidera Dicent : Tu Regere Imperio Populos, Romane, Memento (hae tibi erunt Artes), Pacique Imponere 6th, 2024

### **CHEMISTRY COLLIGATIVE PROPERTIES WORKSHEET Colligative ...**

Colligative Properties Are Properties That Depend Only Upon The Number Of Solute Atoms, Ions, Or Molecules In A Solution And Not On The Nature Of Those Atoms, Ions Or Molecules. Freezing Point Depression And Boiling Point Elevation Are Examples Of Colligative Properties. 7th, 2024

### **Colligative Properties- Page 1 Lecture 4: Colligative ...**

B Raises Boiling Point Of Water From 100 °C To 100.64 °C •  $\Delta T_f$  Lowers Freezing Point Of Water From 0 °C To -2.32 °C •  $\pi$  Raises Osmotic Pressure From 1 atm To 28.6 atm While The Temperature Was Changing By A Degree 9th, 2024

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## 13 Ions In Aqueous Solutions And Colligative Properties

Ions In Aqueous Solutions And Colligative Properties MIXED REVIEW SHORT ANSWER Answer The Following Questions In The Space Provided. 1. Match The Four Compounds On The Right To Their Descriptions On The Left. B An Ionic Compound That Is Quite Soluble In Water (a) HCl C An Ionic Compound That Is Not Very Soluble In Water (b) NaNO<sub>3</sub> 1th, 2024

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### **Quiz: HW02 - Colligative Properties And Solubility Equilibria**

What Will Be The Freezing Point Of A Solution Of 8 Moles Of Sodium Dichromate ( $\text{Na}_2\text{Cr}_2\text{O}_7$ ) Dissolved In 16 Kg Of Water? Use The Following Values:  $K_f = 1.86^\circ\text{C/m}$   $K_b = 0.512^\circ\text{C/m}$  Question 8 1 Pts 20.90 Torr 28.10 Torr The Vapor Pressure Would Not Change As Nap 9th, 2024

### **WORKSHEET:SOLUTIONS AND COLLIGATIVE PROPERTIES SET A**

Answer:  $[\text{K}^+] = 0.71\text{ M}$ ,  $[\text{SO}_4^{2-}] = 0.24\text{ M}$ ,  $[\text{PO}_4^{3-}] = 0.074\text{ M}$ . 4. Calculate The Concentration Of Each Ion And The Mass Of Any Precipitate When A 0.300 Mole Of Aluminum Hydroxide Is Added To 50.0 mL Of 2.5 M Nitric Acid Solution ( Assume That There Is No Volume Change Upon 8th, 2024

### **Colligative Properties And Concentration Calculations**

19. You Have Three Beakers, Each Containing A Solution Of Glucose. The First Beaker Contains 38.9 mL Of 1.62 M Glucose. The Second Beaker Contains 0.0248 L Of 1.38 M Glucose. The Third Beaker Contains 4.39 mL Of 1.08 M Glucose. If You Were To Mix All Three Beaker Together, What Wou 3th, 2024

### **RULE 1. INTERPRETATION 1 RULE 2. COVERAGE 1 RULE 3 ...**

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