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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g

Theory yield = $5.95\text{g KClO}_4 \times 1\text{mol} / 122.55\text{g} \times 2 \times 16\text{g} = 3.11\text{g O}_2$

$1.45\text{g} / 3.11\text{g} \times 100\% = 46.6\%$

34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure FeCO_3 Apr 9th, 2024.

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