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ENTHALPY AND ENTROPY OF A BORAX SOLUTION

Since Borate Ion Is A Base, Its Concentration Can Easily Be Determined By A Simple Acid-base Titration. The Endpoint Is Signaled By The Color Change Of Bromocresol Green Indicator. (7) $B_4O_5(OH)_4^{2-} + 2H^+ + 3H_2O \rightarrow 4B(OH)_3$ Example: A Student Prepared 5.00-mL Of Borax Solution At 59.2°C. He Diluted The Solution In A 2th, 2024

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