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Answers To Practice Test Questions 9 Polarity ...(see Lewis Structure). 7. Both Tetrahalides Are Tetrahedral, Nonpolar Compounds, So The Only Intermolecular

Forces Present Are Induced Dipole-induced Dipole Forces • CCl_4 Is Smaller Than Cl_4 . Therefore, Cl_4 Has A Much Higher Polarizability And Has Much Stronger Induced Dipole-induced Dipole Forces. As A Consequence, Cl_4 Has A Higher z ,
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