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**Colligative Properties- Page 1 Lecture 4: Colligative ...**

B Raises Boiling Point Of Water From 100 O C To

100.64o C • ΔT F Lowers Freezing Point Of Water From 0o C To -2.32o C • π Raises Osmotic Pressure From 1 Atm To 28.6 Atm While The Temperature Was Changing By A Degre Apr 1th, 2024

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**12.3 Colligative Properties - REMONDINI**
Colligative Properties For Molecular Solute Vs. Ionic Solute. Example: 1.0 M Of Sugar & 1.0 M Of Salt Show Different ΔT B And ΔT F Even Though The Molality Concentrations Of The Solute Are Identical. Note That; ΔT = M • K F,b Where K Is Only A Function Of The Solvent !!! Reason For The Difference In Colligative Properties Must Be Based On ... Mar 1th, 2024

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