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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  (1)  $\Delta H^\circ = -92 \text{ kJ mol}^{-1}$   $\Delta S^\circ = -192 \text{ J mol}^{-1} \text{ K}^{-1}$  5th, 2024.  
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