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A)  $[\text{Co}(\text{H}_2\text{O})_6]^{2+} + \text{Co}^{+2}$  E)  $[\text{FeO}_4]^{2-} + \text{Fe}^{+6}$  I)  $\text{K}_2[\text{CoCl}_4] + \text{Co}^{+2}$  B)  $[\text{CrCl}_6]^{3-} + \text{Cr}^{+3}$  F)  $[\text{Mn}(\text{CN})_6]^{4-} + \text{Mn}^{+2}$  J)  $\text{K}_3[\text{AuF}_6] + \text{Au}^{+3}$  C)  $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2 + \text{Co}^{+2}$  G)  $[\text{Ni}(\text{CO})_4] + \text{Ni}^0$  K)  $(\text{NH}_4)_2[\text{IrCl}_6] + \text{Ir}^{+4}$  D)  $[\text{Co}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2 + \text{Co}^{+3}$  H)  $[\text{Ni}(\text{EDTA})]^{2-} + \text{Ni}^{+2}$  L)  $\text{Na}[\text{Mn}(\text{CO})_5] + \text{Mn}^{-1}$  TASK 2 - Writing Half Equations A)  $2\text{I}^- \rightarrow \text{I}_2 + 2\text{e}^-$  - B ... Mar 17th, 2024

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Metal Carbonate  $\rightarrow$  Metal Oxide + Carbon Dioxide (on Heating)  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. Jan 26th, 2024

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