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3 Equivalent Bonds Oriented At 90° To Each Other, And The S Electron Would Form A Bond Of A Different Type And Orientation From The Other Three. No Such Compound Exists. The Simplest Hydrocarbon -methane (CH₄)- Is Known To Have Tetrahedral Geometry, Where The Four C-H Bonds Are All Equivalent And Positioned At 109.5° Angles To Each Other. Jan 6th, 2024

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