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Think Of As Hydrogen, Is Called Hydrogen-1 And Is Stable. The Other Two Isotopes Are Called Hydrogen-2 (deuterium) And Hydrogen-3 (tritium). Deuterium 1th, 2024

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The Binding Energy Per Nucleon Dictates Results When Nuclides Break Apart (fission) Or Fuse Together (fusion) (keep In Mind That ... (two Protons And Two Neutrons) ! "= 2 4He+2 ... Or An Anti-neutrino ( ) Neutrinos Have No Charge, Spin 1/2, And Mass ~ 0.1 -

1 EV (?) The Beta Particle 1th, 2024

CHAPTER 1 Radioactive Decay & Decay Modes
This Equation Represents A First-order Reaction And In The Present Case A Mononuclear Reaction. Radioactive Decay Is Only Possible If ΔE > 0. ΔE Can Be Determined By Comparison Of The Masses. According To The Relation By Einstein (E = Mc2): (EQ 3) By Calculation Of ΔE It C 4th, 2024

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0e Kt; ð1Þ Where N(t) Is The Number Of Atoms Remaining At Time T, N 0 = N(0) Is The Initial Number Of Atoms, And K = Ln(2)/T 1/2 Is A Constant Determined By The Nuclear Half-life T 1/2. However, Recent Work Has Uncovered Evidence Of An Unexpected Periodicity In Several Nuclear Decays [2 1th, 2024

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