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139 kJ/mol (3) 140 kJ/mol (4) 150 kJ/mol Ans. (Sol. $\text{C}_2\text{H}_4 + \text{H}_2 \rightarrow \text{C}_2\text{H}_6$ | $\Delta H = -120$
 kJ/mol) Ph $\text{C}_6\text{H}_5\text{OH} + \text{ICH}_3 \rightarrow \text{PhC}_6\text{H}_4\text{OCH}_3 + \text{HI}$ 140. Which is correct order of reactivity? (1) $\text{CH}_3\text{COCH}_3 > \text{CH}_3\text{CHO} > \text{CH}_3\text{C}_6\text{H}_4\text{CHO}$ (2) $\text{CH}_3\text{COCH}_3 > \text{CH}_3\text{C}_6\text{H}_4\text{CHO} > \text{CH}_3\text{CHO}$ Apr
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 $\rightarrow \text{C}_3\text{H}_8\text{(g)} + \text{O}_2\text{(g)}$ = Water Gas Or Synthesis Gas Temporary Hardness Of Water Is Due To
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