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10. Write The Molecular And Net Ionic Equations For The Following Aqueous Reactions. If No Reaction Occurs, Write NR. A) Ba(NO3)2 + K2SO4 B) Mg(NO3)2 + K2SO4 C) CaCl2 + Al(NO3)3 D) NiCl2 + NaBr E) Na2CO3 + MgBr2. 11. Write The Net Ionic Equation For

Each Of The Following. Use The General Rules For Solubility (Bauer Table 5.3, P 177). 8th, 2024

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System? A. HCl, NaCl B. HNO 3, NaNO 3 C. H 3 PO 4, H
2 PO 4-D. H 2 SO 4, CH 3 COOH E. NH 3, NaOH 2.
Which, If Any, Of The Following Aqueous Mixtures
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