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What Would Be The Weight Of This Maximum Yield? To Answer This, We Need To Convert Our Mole Value For H<sub>2</sub>S To A Mass Value. Remember That  $M = nM_w$  We Can Specify  $M_w = (2 \times 1,008) + 32.06 = 34.08 \text{ g/mol}$  Proto  $M = 3.5 \times 34.08 = 119.3 \text{ grams}$  H<sub>2</sub>S Could Be Produced. III. If You Would Like To Produce 3.00 Kg H<sub>2</sub>S  
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Model 2: The Mole Concept 1 Dozen Items = 12 Dozen Items = 1 Egg Sandwich = 1



Mole = Key Questions C 12 Items 1 Gross Of Items 3 Eggs And 2 Slices Of Bread  
 $6.02 \times 10^{23}$  items (Avogadro's Number) Use Dimensional Analysis ONLY For All  
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500 Mol Li 2O 29. 0 G H 2O 2) How Many Grams Are In 4. How Many Grams Of Cal.  
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× 10<sup>23</sup> molecules) 1 Mole 1 Mole Pogil Relative Mass And The Mole Answers Tip  
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