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Molar Gas Constant  $R = 8.314\ 472(15)\ \text{J K}^{-1}\ \text{mol}^{-1}$  Molar Volume Of An Ideal Gas ( $P = 101.325\ \text{kPa}$ ,  $T = 0\ ^\circ\text{C}$ )  $V_M = 22.412\ 996(39)\ \text{dm}^3\ \text{mol}^{-1}$  Standard Partial Molar Enthalpy  $H_B^\circ\ \text{J mol}^{-1}$  Standard Partial Molar En Apr 2th, 2024

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