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A. The Number Of Atoms In 2.0 Mole Al Is 2.0 Mole Al X 6.02x1023 Al Atoms = 1.2 X 1024Al Atoms 1 Mole Al Avogadro's Number B. The Number Of Moles Of S In 1.8 X 1024 Atoms S Is 1.8 X 1024 Atoms S X 1 Mole S 6.02 X 1023 S Atoms Avogadro's Number = 3.0moles Of S Atoms Solution 7th, 2024

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