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Experiment 10 Stoichiometry- Gravimetric Analysis

Experiment 10 Stoichiometry- Gravimetric Analysis 10- 2 Materials And Equipment Solid Sodium Carbonate (Na_2CO_3), Sodium Bicarbonate (NaHCO_3), 6.0 M Hydrochloric Acid (HCl), Balance, Evaporating Dish, Watch Glass (to Fit As A Cover For The Evaporating Dish), Stand And Ring Clamp, Wire Gauze, Dropper Pipette, Stir Rod And Bunsen Burner. Part A 1. Jan 1th, 2024

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Gravimetric Analysis Of An Unknown Sulfate Page 3 Of 4 Questions 1. Suppose An Unknown Metal Sulfate Is Found To Be 72.07 % SO₄²⁻. Assuming The Charge On The Metal Cation Is +3, Determine The Identity Of The Cation. 2. The Unknown Metal Sulfates Are Hygroscopic And Will Absorb Water From Air. The Unknowns Apr 12th, 2024

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Example Of Calculations Used In Gravimetric Analysis: 0.7246 G Of An Unknown Was Analyzed By Gravimetric Analysis Using A 26.6002g Crucible Filter. Calcium Oxalate Monohydrate Was Precipitated, Washed And Dried, And The Resulting Precipitate Plus Crucible Was 28.1023 G. What Is The % CaO In The Unknown Sample? Jan 7th, 2024

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