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A) [Co(H₂O)₆]²⁺ Co + 2 E) [FeO]₄²⁻ Fe + 6 I) K₂[CoCl₄] Co + 2 B) [CrCl₆]³⁻ Cr + 3 F) [Mn(CN)₆]⁴⁻ Mn + 2 J) K₃[AuF₆] Au + 3 C) [Co(NH₃)₆]Cl₂ Co + 2 G) [Ni(CO)₄] Ni 0 K) (NH₄)₂[IrCl₆] Ir + 4 D) [Co(NH₃)₅Cl]Cl₂ Co + 3 H) [Ni(EDTA)]²⁻ Ni + 2 L) Na[Mn(CO)₅]

Mn -1 TASK 2 – Writing Half Equations A) $2I^- \rightarrow I_2 + 2e^-$
– B ... 3th, 2024

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It Burns ... 1th, 2024

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Metal Carbonate \rightarrow Metal Oxide + Carbon Dioxide (on Heating)
 $CaCO_3 \rightarrow CaO + CO_2$ TASK 4 – WRITING
BALANCED EQUATIONS 1) Balance The Following
Equations. 11th, 2024

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Calculate T 8th, 2024

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(A2 BBoF 23) 4 Kc For The Equilibrium Below Is 7.50 At
50 °C. If 50.0 G Of Ethanol Is Mixed With 50.0 G Of
Propanoic Acid, What Mass Of Ethyl Propanoate Will Be
Formed At Equilibrium? $\text{CH}_3\text{CH}_2\text{OH(l)} + \text{CH}_3\text{CH}_2\text{COOH(l)} \rightleftharpoons \text{CH}_3\text{CH}_2\text{COOCH}_2\text{CH}_3\text{(l)} + \text{H}_2\text{O(l)}$ ((((((
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An Experiment To Find The Activation Energy For The
Following Reaction: $2 \text{MnO}_4 \dots$ 8th, 2024

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Reaction. The ... 9th, 2024

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For Some Addition Reactions Of Alkenes. You Should
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