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Groups: 1) Pure Substances, Such As Elements And Compounds, And 2) Mixtures. Unlike The Components Of A Compound, The Components Of A Mixture Can Be Separated By Physical Changes. In Addition, The Amount Of Each Component In A Mixture Is Not Fixed. What Is A Heterogeneous Mixture? Exam 4th, 2024

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That Electrolysis Of A Dilute Aqueous Solution Of An Ionic Compound (magnesium Sulfate) Produces A Gas At Both Electrodes And A Basic Solution At The 4th, 2024

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