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 $2 \rightarrow$  CaSiO 3 B. 3Fe 2 O 3 + CO  $\rightarrow$  2Fe 3 O 4 + CO 2 C. FeO +SiO 2  $\rightarrow$  FeSiO 3 D. FeO  $\rightarrow$  Fe + 1 2 O 2 A. A B. D C. C And D D. A And D Answer: C Solution: In Metallurgy Of Iron, CaO Is Used As Flux Which Is Used To Remove The Impurities Of SiO 2 , CaO + SiO 2  $\rightarrow$  CaSiO 3 . Also Here Fe 2 O 3 Is Reduced By CO To Fe 3 O 4 Which Is Further Reduced To ... Mar 7th, 2024

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