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A)  $[\text{Co}(\text{H}_2\text{O})_6]^{2+}$  Co +2 E)  $[\text{FeO}_4]^{2-}$  Fe +6 I)  $\text{K}_2[\text{CoCl}_4]$  Co +2 B)  $[\text{CrCl}_6]^{3-}$  Cr +3 F)  $[\text{Mn}(\text{CN})_6]^{4-}$  Mn +2 J)  $\text{K}_3[\text{AuF}_6]$  Au +3 C)  $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$  Co +2 G)  $[\text{Ni}(\text{CO})_4]$  Ni 0 K)  $(\text{NH}_4)_2[\text{IrCl}_6]$  Ir +4 D)  $[\text{Co}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2$  Co +3 H)  $[\text{Ni}(\text{EDTA})]^{2-}$  Ni +2 L)  $\text{Na}[\text{Mn}(\text{CO})_5]$  Mn -1 TASK 2 - Writing Half Equations A)  $2\text{I}^- \rightarrow \text{I}_2 + 2\text{e}^-$  - B ... Feb 21th, 2024

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Metal Carbonate  $\rightarrow$  Metal Oxide + Carbon Dioxide (on Heating)  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. Apr 20th, 2024

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