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## **Hydrolysis Of Uniunivalent Salts In Aqueous Solutions. II ...**

Weak Acids, HA, Where A<sup>-</sup> Is The Anion: (2) And Weak Bases, BOH, Where B<sup>+</sup> Is The Cation: These Weak Electrolyte Cases, Treated By An Ionization Constant, Differ From The Strong Electrolyte Cases Whose Essentially Complete Dissociation Makes It Possible To Take The Concentration Feb 9th, 2024

## **HYDROLYSIS OF SALTS**

Salt Solutions May Be Acidic, Basic, Or Neutral, Depending On The Original Acid And Base That Formed The Salt. Strong Acid + Strong Base Neutral Salt Strong Acid + Weak Base Acidic Salt ... 1. Complete The Following Hydrolysis Reactions For These

Ions. F-(aq) + H<sub>2</sub> Feb 7th, 2024

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