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18. Write The Abbreviated (noble Gas) Electron Configuration For The Following

Atoms: A. Silver B. Copper C. Tungsten D. Titanium 19. Define Ground State. 20. Define Excited State. 21. How Do Atoms Jump From Ground State To Excited State And Back? (Remember Our Flame Test L 5th, 2024

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5. List The Seven Diatomic Molecules. Chapter 9: Chemical Names & Formulas 1. Determine Whether The Following Compounds Are Ionic Or Covalent, And Then Name Them. CaSO_4 , N_2O_4 , NH_4NO_3 , CaS , Li_3PO_4 , NH_3 , $\text{Cu}(\text{OH})_2$. 2. Complete The Following Table And Name Each Ionic Compound Formed. + 26th,

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C. Ag-107, Au-197 And Ar-40 11. List The Order That You Fill The Energy Levels And Orbitals For Electron Configuration. 12. Fill In The Orbital Diagram Below For

Tungsten And Then Write The Full Electron Configuration Below. W: 13. Which Of The Following Is An Incorrect Orbital Diagram? Why? 17th, 2024

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A. It Gains Electrons B. It Gives Up Electrons C. It Does Not Change Its Number Of Electrons D. Calcium Does Not Obey The Octet Rule 3. What Is The Formula When Potassium And Sulfur Combine? K₂S 4. What Is The F 13th, 2024

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