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87 The Total Number Of Moles, N , Is The Sum Of The Number Of Moles Of All The Components Of The Mixture: $N = N_1 + N_2 + \dots$ It Follows That The Sum Of All The Mole Fractions Of The Mixture Must Also Equal 1. $X_1 + X_2 + \dots = 1$ The Mass Of Component I In A Mixture Is The Product Of The Number Of Moles Of I And Its Molecular Weight, M_i . The Mass Of The Mixture Is Therefore The Sum, $M = N_1M_1 + N_2M_2 + \dots$ Jan 28th, 2024

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The Time That Elapses From The Start Of Fuel Mar 29th, 2024

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Fuel Gas Code ANSI Z223.1, Or - Sections 7.2, 7.3, Or 7.4 Of CAN/CGA B149,
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Temperature: • Category I, Mar 30th, 2024

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The Flow Of Air Or Gas Through An Orifice Can Be Determined By The Formula $Q = 1658.5 \times A \times C_d \times \sqrt{H \times G}$ Where Q=flow, Cfh A=area Of The Orifice, Sq. In. (see Pages 57 & 58) C_d =discharge Coefficient Of The Orifice (see Above) H=pressure Drop Across The Orifice,"w.c. G=specific Gravity Of The Gas, Based On Standard Air At 1.0 (see Pages 19, 20, & 22 ... Feb 2th, 2024

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