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Tert-butoxide ion is very bulky, or sterically hindered, and is less capable of interacting with the solvent. The ethoxide ion is not as sterically hindered so it can accommodate more solvent interactions. As a result, ethoxide is better solvated and is therefore more stable than tert-butoxide (Figure 3.7).

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ORGANIC CHEMISTRY LABORATORY EXPERIMENTS FOR ORGANIC ...ORGANIC CHEMISTRY 121 EXPERIMENT 1 SYNTHESIS OF ASPIRIN FROM SALICYLIC ACID Aspirin is one of the oldest and most common drugs in use today. It is both an analgesic (pain killer) and antipyretic (reduces fever). One method of preparation is to react salicylic acid (1) with acetic anhydride (2) and a trace amount of acid (equation 1).  $\text{OH COOH O} \dots$  9th, 2024 Advanced Organic Chemistry – Organic Reactivity And Total ...Reduction,

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