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Section 3 Classifying Chemical Reactions
Decomposition Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or

More Substances. $AB \rightarrow A + B$ Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity. Feb 5th, 2024.

Classifying Chemical Reactions
Classifying Chemical Reactions Types Of Chemical Reactions: 1. Synthesis 2.

Decomposition 3. Combustion 4. Single-Replacement 5. Double-replacement *Classifying How Atoms Rearrange During A Chemical Reaction. December 12, 2014

1. Synthesis Reaction: 2 Or More Substances React To Produce A Single Product $A + B \rightarrow AB$ Mar 14th, 2024

Section 8-3: Classifying Chemical

Reactions Section 8-3: Classifying Chemical Reactions

Reaction Type General Format Characteristics Of

Reactants Characteristics Of Products Example (s)

Decomposition $AB \rightarrow A + B$ Only One Reactant Two Or

More Simpler Products $CuCO_3 (s) \rightarrow CuO + CO_2 (g)$ $2 H_2 O (l) \rightarrow 2 H_2 (g) + O_2 (g)$ Synthesis AKA: Combination

$O (l) \rightarrow A + B$ AB 2 Two Elements Or Two Simple Feb 24th, 2024

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TOPIC 33 - CLASSIFYING CHEMICAL REACTIONS PART

1 Specific Types Of Chemical Reactions - Synthesis,

Decomposition And Combustion In Air - Will Be

Explored In This Topic While The Remaining Three -

Single Replacement , Double Replacement And
Neutralization – Will Be Discussed Mar 13th, 2024

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