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A) $[\text{Co}(\text{H}_2\text{O})_6]^{2+} + \text{Co} + 2 \text{E}) [\text{FeO}_4]^{2-} - \text{Fe} + 6 \text{I}) \text{K}$
 $2[\text{CoCl}_4] + \text{Co} + 2 \text{B}) [\text{CrCl}_6]^{3-} - \text{Cr} + 3 \text{F}) [\text{Mn}(\text{CN})_6]^{4-}$
 $\text{Mn} + 2 \text{J}) \text{K} 3[\text{AuF}_6] + \text{Au} + 3 \text{C}) [\text{Co}(\text{NH}_3)_6]\text{Cl}_2 + \text{Co} + 2$
G) $[\text{Ni}(\text{CO})_4] + \text{Ni} 0 \text{K}) (\text{NH}_4)_2[\text{IrCl}_6] + \text{Ir} + 4 \text{D}) [\text{Co}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2 + \text{Co} + 3 \text{H}) [\text{Ni}(\text{EDTA})]^{2-} - \text{Ni} + 2 \text{L}) \text{Na}[\text{Mn}(\text{CO})_5]$
 Mn^{-1} TASK 2 – Writing Half Equations A) $2\text{I}^- \rightarrow \text{I}_2 + 2\text{e}^-$
– B ... Jan 24th, 2024

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Metal Carbonate \rightarrow Metal Oxide + Carbon Dioxide (on Heating) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ TASK 4 – WRITING BALANCED EQUATIONS 1) Balance The Following Equations. Feb 23th, 2024

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