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Show Your Work, Including Proper Units, To Earn Full  
Credit. 1. Silver And Nitric Acid React According To The  
Following Balanced Equation:  $3 \text{Ag(s)} + 4 \text{HNO}_3 \text{(aq)} \rightarrow 3 \text{AgNO}_3 \text{(aq)} + 2 \text{H}_2\text{O(l)} + \text{NO} \dots$  9th, 2024

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Stoichiometry: Mixed Problems (KEY) 1)  $N_2 + 3H_2 \rightarrow 2NH_3$   
What Volume Of  $NH_3$  At STP Is Produced If 25.0

Of N<sub>2</sub> Is Reacted With An Excess Of H<sub>2</sub>? 3 3 3 2 3 2 2 2  
40.0L NH 1mol NH 22.4 7th, 2024

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Practice Work 53 – Stoichiometry-04 Mixed Stoichiometry Problems General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment. Sorry About The Lack Of Format. I'm In A Time Crunch. 123.88 G/mol

70.90 G/mol 137.32 ... 2th, 2024

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24 2 2 23 2 2 2 2 4.63 X 10 molecules | 1 Mol | 6.02 X  
10 Molecules | 1 Mol Cl 1mol 71 G Cl Cl X 546 G Cl 10)  
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Cu File Size: 108KB Page Count: 4 5th, 2024

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Directions: Solve Each Of The Following Problems.

Show Your Work, Including Proper Units 1. \_\_\_ CaCl<sub>2</sub> + \_\_\_ AgNO<sub>3</sub> \_\_\_ Ca(NO<sub>3</sub>)<sub>2</sub> + \_\_\_ AgCl How Many Grams Of Silver Chloride Are Produced When 45 G Of Calcium Chloride React With Excess Silver ... 13th, 2024

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A Practice Problem On Stoichiometry -- ANSWERS

Consider The Reaction Represented By The Equation Below For All Parts Of This Question:  $2 \text{BrCl}_3 \rightarrow 3 \text{Cl}_2 + \text{Br}_2$  (a) If 4 Moles Of  $\text{BrCl}_3$  Reacts According To The Equation, How Many Moles Of  $\text{Cl}_2$  Will Be Formed? How Many Moles Of  $\text{Br}_2$  Will Be Formed? Answers: 6 Mol  $\text{Cl}_2$  And 2 Mol  $\text{Br}_2$  Setups/Work: 15th, 2024

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