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Stoichiometry Problem Sheet 1 Answers Stoichiometry Volume Problems Worksheet Answers Stoichiometry: Mixed Problems (KEY) 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  What Volume Of  $\text{NH}_3$  At STP Is Produced If 25.0 Of  $\text{N}_2$  Is Reacted With An Excess Of  $\text{H}_2$ ? 3 3 3 2 3

2 2 2 40.0L  $\text{NH}_3$  1mol  $\text{NH}_3$  22.4 24th, 2024 Stoichiometry Problem Sheet 1 Answers File Type Online Library Stoichiometry

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33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$  Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1\text{mol} \times 122.55\text{g} \times 2\text{O}_2 \times 32.00\text{g} \times 1\text{mol} = 3.11\text{g}$

$\text{O}_2$  %Yield =  $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$  Percent purity = 62.8%? g pure  $\text{FeCO}_3$  3th,

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