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What Number Of Moles Of  $O_2$  Is Needed To Produce 14.2  
Grams Of  $P_4O_{10}$  From  $P$ ? (Molar Mass  $P_4O_{10} = 284$ )  
(A) 0.0500 Mole (B) 0.0625 Mole (C) 0.125 Mole Apr 11th,  
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Potassium Chlorate (KClO<sub>3</sub>) Is Heated, It Decomposes To Form Gases KCl And Oxygen (O<sub>2</sub>.)  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  C. May 13th, 2024

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