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Solution  
 $M_{Ag} = 11.1 \text{ G}$   
 $M_{Alloy} = 12.0 \text{ G}$   
 $C_{Ag} = \frac{11.1 \text{ G}}{12.0 \text{ G}} \times 100\% = 92.5\%$   
The Ring Is 92.5 % W/W Silver. Very Low Concentrations In Studies Of Solutions In The Environment, We Often ...  
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