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Crystal. Answer: The Scheme Below Represents Born-Haber Cycle For BaO Crystal.
Step 1. Sublimation Of Solid Ba: $\text{Ba}(s) \rightarrow \text{Ba}(g)$, $\Delta H_1 = 180.0 \text{ KJ/mol}$ During This
Process, The Change In Enthalpy Is Positive, Energy Is Required To Change The
State Of Barium From Solid To Gaseous (to Break Metal Lattice). Step 2. Jan 19th,
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