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Grams Per Mole Of CO" So We Do Not Need To
Recalculate That. Two Moles Of Butane ($2 \text{ Mol} \times 58$
 $\text{G/mol} = 116 \text{ G}$) Yields 8 Moles Of Carbon Dioxide (8
 $\text{Mol} \times 44 \text{ G/mol} = 352 \text{ G CO}_2$) , So We Can Set Up The
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