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C. Ag-107, Au-197 And Ar-40 11. List The Order That You Fill The Energy Levels And Orbitals For Electron Configuration. 12. Fill In The Orbital Diagram Below For Tungsten And Then Write The Full Electron Configuration Below. W: 13. Which Of The Following Is An Incorrect Orbital Diagram? Why? Feb 20th, 2024

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18. Write The Abbreviated (noble Gas) Electron Configuration For The Following Atoms: A. Silver B. Copper C. Tungsten D. Titanium 19. Define Ground State. 20. Define Excited State. 21. How Do Atoms Jump From Ground State To Excited State And Back? (Remember Our Flame Test L Mar 5th, 2024

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2. How Does Calcium Obey The Octet Rule When Reacting To Form Compounds? A. It Gains Electrons B. It Gives Up Electrons C. It Does Not Change Its Number Of Electrons D. Calcium Does Not Obey The Octet Rule 3. What Is The Formula When Potassium And Sulfur Combine? K<sub>2</sub>S 4. What Is The F Jan 19th, 2024

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9. Find A Unit Vector In  $R^2$  That Is Orthogonal To  $\begin{pmatrix} 1 \\ 2 \end{pmatrix}$  .  
Solution: We Want To Find A Vector  $\sim v = \begin{pmatrix} v_1 \\ v_2 \end{pmatrix}$  Such That  $\sim v \cdot \begin{pmatrix} 1 \\ 2 \end{pmatrix} = 0$ . Evaluating This Dot Product Gives The



Equation  $v_1 + 2v_2 = 0$  So  $v_1 = -2v_2$ : Thus, Any Vector Of The Form  $\vec{u} = v_2 \begin{pmatrix} -2 \\ 1 \end{pmatrix}$  Is Orthogonal To  $\begin{pmatrix} 1 \\ 2 \end{pmatrix}$ .  
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