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(b) He₂ Is Not Stable But He₂⁺ Is Expected To Exist. (c) Bond Strength Of N₂ Is Maximum Amongst The Homonuclear Diatomic Molecules Belonging To The Second Period. Thus, Dinitrogen (N₂) Molecule Contains Triple Bond And No Other Molecule Of Second Period Have More Than Double Bond. Hence, Bond Strength Of N₂ Is Maximum Amongst The Homonuclear Jan 4th, 2024

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Aug 25, 2014 · 5) One Mole Of Nitrogen Gas Contains (2) × (6.022 × 10²³) Nitrogen Atoms. 5) 6) One Mole Of Chlorine Gas Has A Mass Of 35.45 Grams. 6) 7) One Mole Of I₂ Has More Atoms In It Than One Mole Of Na. 7) 8) Six Grams Of Carbon Contains 3.008 × 10²³ Atoms. 8) 9) Two Mole(s) Of CO₂ Gas Contains 1 Mole Of Carbon Mar 11th, 2024

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