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$+ 2\text{H}_2\text{O} \rightarrow 2\text{LiOH} + \text{H}_2$  3. Decomposition  $2\text{AlCl}_3 \rightarrow 2\text{Al} +$

$3\text{Cl}_2$  Mar 4th, 2024

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Compounds Are Formed When Atoms Are Bonded

Together. 9 Breaking A Chemical Bond Is An

Endothermic Process. 9 Forming A Apr 8th, 2024.

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4.1 Explain The Formation Of A Chemical Bond. Answer  
: The Attractive Force Which Holds Various  
Constituents (atoms, Ions, Etc.) Together In Different  
Chemical Species Is Called A . Jan 17th, 2024.  
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Chemical ... (b)  $\text{He}_2$  Is Not Stable But  $\text{He}^{+2}$  Is Expected  
To Exist. (c) Bond Strength Of  $\text{N}_2$  Is Maximum Amongst  
The Homonuclear Diatomic Molecules Belonging To  
The Second Period. Thus, Dinitrogen ( $\text{N}_2$ ) Molecule  
Contains Triple Bond And No Other Molecule Of Second  
Period Have More Than Double Bond. Hence, Bond  
Strength Of  $\text{N}_2$  Is Maximum Amongst The Homonuclear  
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Exam Accelerated Chemistry Study Sheet Chap 06 Chemical ... Aug 25, 2014 · 5) One Mole Of Nitrogen Gas Contains  $(2) \times (6.022 \times 10^{23})$  Nitrogen Atoms. 5) 6) One Mole Of Chlorine Gas Has A Mass Of 35.45 Grams. 6) 7) One Mole Of  $I_2$  Has More Atoms In It Than One Mole Of Na. 7) 8) Six Grams Of Carbon Contains  $3.008 \times 10^{23}$  Atoms. 8) 9) Two Mole(s) Of  $CO_2$  Gas Contains 1 Mole Of Carbon Apr 5th, 2024  
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