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CHEMISTRY 9701/01 1 Hour 10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established.  $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$  What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The  $V^{2+}$  Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Apr 26th, 2024.

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CHEMISTRY 9701/04(c) (i) Write A Chemical

Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s). First Ionisation Energy Of Silver = +731kJmol<sup>-1</sup> Electron Affinity Of Bromine = 325kJmol<sup>-1</sup> Enthalpy Change Of Atomisation Of Silver = +285kJmol<sup>-1</sup> Enthalpy Change Of Atomisation Of Bromine = +112kJmol<sup>-1</sup> ... Jan 9th, 2024.

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