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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established.  $V^{2+} + H_2O \rightleftharpoons V^{3+} + H^+$   
 $2 H_2O \rightleftharpoons H_3O^+ + OH^-$  What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The  $V^{2+}$  Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Mar 8th, 2024

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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s). First Ionisation Energy Of Silver =  $+731 \text{ kJ mol}^{-1}$  Electron Affinity Of Bromine =  $325 \text{ kJ mol}^{-1}$  Enthalpy Change Of Atomisation Of Silver =  $+285 \text{ kJ mol}^{-1}$  Enthalpy Change Of Atomisation Of Bromine =  $+112 \text{ kJ mol}^{-1}$  ... Feb 2th, 2024

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