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Name: AP Chemistry Period: Date: Dr. Mandes The Following Questions Represent Potential Types Of Quiz Questions. Please Answer Each Question Completely And Thoroughly. The Solutions Will Be Posted On-line On Monday. 5. Please Do #18 In Chapter 12 Of Your Text. A. This Increases The  $[H_2]$ , Which Will Increase The Rate, But Has No Effect On  $k$ . B. Due To Arrhenius Equation, Changing Temperature ... Jan 23th, 2024

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$k_1 [D] [B] [Y] = k_2 [D] [B] [Y]$  Equation 5 In This Equation  $k$  Cancels Out.  $[D]_1 = [D]_2$  Because We Are Starting Off With The Same Initial Concentration Of  $A$  In Each Trial. The Order Of Species  $D$  Which Is  $X$  Is Also Constant. Canceling Terms We Have Left:  $Rate_1 = Rate_2 = k [B] [Y]$  Equation 6 Consider The Following Chemical Reaction: Apr 23th, 2024

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The Following Graph Shows The Kinetics Curves For The Reaction Of Oxygen With Hydrogen To Form Water:  $O_2(g) + 2H_2(g) \rightarrow 2H_2O(g)$ . Which Curve Is Hydrogen? A. The Dashed Curve B. The Gray Curve C. The Black Curve D. Either The Gray Or The Black Curve E. Any Of These Curves Could Be Hydrogen. 3 Apr 27th, 2024

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