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67 PREPARATION AND TESTING OF BUFFER SOLUTIONS. P. URPOSE. The Purpose Of The Laboratory Investigation Is To Experimentally Determine (1) pK_a (and Thus K_a) Of The Acid In A Buffer And Thus The Buffer Range, (2) Investigate The Buffer

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Section 19.1. Acid-Base Buffer Solutions

If We Add CH_3COO^- -SHIFT Net Effect: $[\text{H}^+]$ Decreases pH Increases Also, $[\text{OH}^-]$ Increases POH Decreases I.e., a Weak Acid And Its Conjugate Base Or A Weak Base And Its Conjugate Acid. 19-2 Example: Addition Of NH_4Cl To NH_3 Mar 8th, 2024

A Simple And Universal Method For Making Up Buffer Solutions

TBoyer, R F (1986) 'Modern Experimental Biochemistry', Addison- Wesley Publishing Co, Reading, MA 2Bohinski, R C (1983) 'Modern Concepts In Biochemistry', Fourth Edition, Allyn And Bacon, Boston, MA 3Lehninger, A L (1982) 'Principles Of Biochemistry', Worth Publishers, New York Mar 2th, 2024

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