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CHAPTER 4 TEST: Atoms, Atomic Theory And Atomic StructureUse The Diagram To The Right To Answer Question 1. _____1. The Atomic Number Of Carbon Is 6, Which Means That Carbon Atoms Always Have 6 A. Ions B. Protons C. Neutrons D. Valence Electrons _____2. In His Investigations Of Air, Henry Cavendish Discovered A Small Bub Apr 2th, 2024Name: Unit 2: Atomic Structure Atomic Structure And ...Then, Write A Sentence Using The Word But Instead Of Writing The Word, Draw A Picture In Its Place. EXAMPLE Word: Noble Gases Definition: The Elements In Group 8A Or 18 Including Helium, Neon, Argon, Krypton, Xenon, And Radon. Sentence: The Are Called Noble Because They Contain 8 Valence Electrons And Do Not React With Other Elements. Jan 4th, 2024Chapter 4 Atomic Structure Section 4.3 Modern Atomic Theory(pages 113–118) This Section Focuses On The Arrangement And Behavior Of Electrons In Atoms. Reading Strategy (page 113) Sequencing After You Read, Complete The Feb 7th, 2024.

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