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# Thermochemistry, Thermodynamics Thermochemistry

Standard States The Thermodynamic Standard State Of A Substance Is Its Most Stable Pure Form Under Standard Pressure (1 Bar) And Some Speci C Temperature (298 K Unless Otherwise Speci Ed). I For A Pure Substance In The Liquid Or Solid Phase, The Standard State Is Th Mar 26th, 2024

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Bonds), But The Answer Would Be The Same Regardless. You Only Need To Account For Bonds That Participate In The Reaction. 9. Bond Energy Per Bond (kJ Per Bond) H-Cl 427 S-Cl 271 H-S 347 Cl-Cl 243 2 HCl + SCl 2 H 2 S + 2 Cl 2 Reactants Bond Number Of Bonds/Coefficient Bond Energy (each) Total Energy HCl 2 427 854 SCl 2 271 542 Total 1396 Mar 24th, 2024

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11.3: Heat In Changes Of State Enthalpy (H): - The Amount Of Internal Energy At A Specific Pressure And Volume. Energy Involved In Physical Change (Temperature Or Phase Change): Heating Curve: - A Graph Of Temeparture Versus Time As A Substance Is Heated From A Solid Phase To A Gaseous Phase. Mar 7th, 2024

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