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The Metal Calcium, And Predict The Ion Form It Must Become To Be Isoelectric With The Nearest Noble Gas. Answer: Ca:  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2$  Or  $[Ar]4s^2$  The Nearest Noble Gas To Calcium Is  $[Ar]$ . So Calcium Will Typically Lose Two Electrons To Become  $Ca^{2+}$ . Jan 9th, 2024 Chemistry Student Edition - Basic Answer Key Chapter 9 ...Electrons In Different Bonding Pairs Causes Them To Remain As Far Apart As Possible. 2. A. 4 B. 6 C. 3 D. 2 3. Molecular Geometry Only Considers Atoms, While Electron Domain Geometry Includes Lone Pairs 4. Lone Pairs Take Up More Space Than Bonding Pairs 5. A. Bent,