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17.!Explain The Trend In Acidity. Include Key Structures. L The Stronger Acid Has A Lower Energy Conjugate Base. The Cyclopropene Is Least Acidic Because Its Conjugate Base Is Antiaromatic (high Energy, Loop Of P's, 4 π E-). The Linear Diene's Conjugate Base Is Resonance Stabilized, And Therefore Is More Acidic. 3th, 2024

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4. Which Of The Following Ions Does Not Have A Noble-gas Configuration? S2-Y+ P3-Sc3+ Mg2+ 5. Which Of The Following Is Most Likely Possible For An Alkaline Earth Metal? 6. What Is Main Energy Level From Which Electrons Are Lost When 5th, 2024

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9. Name And Draw Structural Formulas For The Cis And Trans Isomers Of 1,2-dimethylcyclopropane. 10. Name And Draw Structural Formulas For The Six Cycloalkanes With The Molecular Formula C 5 H 10. Include Cis-trans Isomers As Well As Constitutional Isomers. 11. Draw Structural Formulas Fo 9th, 2024

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