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2. Q. List Three Subatomic Particles Of An Atom. Compare Them On The Basis Of Relative Mass And Charge In A Tabular Form. Ans: Particle Symbol Charge Mass Electron e^{-} 1.60×10^{-19} C 9.1×10^{-31} Kg Proton $+p$ (H⁺) 1.60×10^{-1} Mar 9th, 2024

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(c) Three Mole (d) Four Moles (4) Which Of The Following Weighs The Most? (a) One G-atom Of Nitrogen (b) One Mole Of Water (c) One Mole Of Sodium (d) One Molecule Of H₂SO₄ (5) 5.0 Litre Of Mar 12th, 2024

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