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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At
The Formation Of Ammonia (NH_3) From Nitrogen And Hydrogen (see Section
Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The
Equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ (1) $H_0 = 92 \text{ kJ Mol}^{-1}$ $S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$ Apr 6th,
2024Chemical Engineering, CM3510: Chemical Reaction ...Grading: Exam 1 15%
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Class Participation 5% Last Year, The Following Grading Scale Was Used: $A \geq 92$ A/B
 ≥ 89 $B \geq 82$ $B/C \geq 80$ $C \geq 71$ $C/D \geq 69$ $D \geq 61$ $F \leq 60$ If Necessary, The Grading
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