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Between The Mass Of A Sample Of An Element And The Number Of Atoms That The  
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Elements. For Example, Sodium Is A Shiny Metal, And Chlorine Is An Irritating Gas.  
But The Compound Formed From These Two Elements, Sodium Chloride, Shows  
Different Characteristics As It Is A Crystalline Solid, Vital For Biological Functions.  
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VIII) Strengths Of Covalent Bonds For Ionic Compounds  $\Delta H_{\text{LE}}$  Is An Indication Of The Strength Of Attraction Of The Ions. For Covalent Bonds The Strength Is Measured As The Bond Dissociation Energy (BDE) Or Bond Enthalpy (BE). Energy Required To Dissociate One Mole Of Bonds | Feb 2th, 2024

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Stoichiometry: Calculations With Chemical Formulas And Equations John D. Bookstaver St. Charles Community College Cottleville, MO Lecture Presentation. ... Mass Of A Compound That Comes From Each Of The Elements In The Compound By Using This Equation:  $\% \text{ Element} = \frac{(\text{number of atoms})(\text{atomic weight})}{(\text{FW of the compound})} \times 100$ . Jan 7th, 2024



## **UNIT (4) CALCULATIONS AND CHEMICAL REACTIONS**

Chemical Reactions. One Approach Is To Classify Reactions Into Four Types: Combination, Decomposition, Single Replacement And Double Replacement Reactions. I) Combination Reactions In A Combination Reaction, Two Or More Substances React To Form A Single Product. The General Form Of This Reaction Is ( $A + B \rightarrow AB$ ). Some Examples Are Shown Below: Apr 18th, 2024

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Chemical Calculations. Measurement In Every Day Life Measurement: ... Divide  $1.14 \div 9.10$  To Get The Final Base  $1.14 \div 9.10 = 0.125$  Eighth, Subtract The Exponents  $(-1) - (-5) = 4$  ... How Many Significant Figures Are In The Following Quantities?  $1.002 \text{ L}$   $36.4 \text{ Cm}$   $6.022 \times 10^{23} \text{ Atoms}$   $2.88790 \times$  Mar 2th, 2024

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With Chemical Formulas And Equations 2 | 1 . 2 ... Percentage Composition  
Example: A Sample Of A Liquid With A Mass Of  $8.657 \text{ G}$  Was ... •Smallest Is  $3.722 \times 10^{-3} \text{ Mole}$  •C = •H = Step 3: Divide All Number Of Moles By The Smallest Jan 20th,

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### **Chapter 3 Calculations With Chemical Formulas And Equations**

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