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Thermodynamics SI Version, 2nd Edition, John Wiley
And Sons, New York, ISBN 0-471-04188-2. Feb 4th,
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1 Chemical Reactions 1.1 Rate Of Reaction And

Dependence On Temperature We Will Once Again Look At The Formation Of Ammonia (NH_3) From Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ (1) $\Delta H^\circ = -92 \text{ kJ mol}^{-1}$ $\Delta S^\circ = -192 \text{ J mol}^{-1} \text{ K}^{-1}$ Feb 6th, 2024

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