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Materials Chemistry: Stoichiometry - Problem Sheet 1
Directions: Solve Each Of The Following Problems.
Show Your Work, Including Proper Units, To Earn Full
Credit. 1. Silver And Nitric Acid React According To The
Following Balanced Equation: $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3 \text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO}$... Jan 3th, 2024

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Answer Key

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Practice Work 53 Stoichiometry-04 Mixed Stoichiometry ...

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Chemistry 11 Stoichiometry Review Worksheet

Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol}$

$122.55 \text{ g} \times \frac{2 \text{ O}_2}{1 \text{ KClO}_4} \times \frac{32.00 \text{ g}}{1 \text{ mol}} = 3.11 \text{ g O}_2$

% Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Purity = 62.8%? pure FeCO_3 Jan 2th, 2024

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Key (ANSWER 386.3g Of LiNO_3) 4) Using The Following

Equation: $\text{Fe}_2\text{O}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 3\text{H}_2\text{O}$. Calculate

How Many Grams Of Iron Can Be Made From 16.5

Grams Of Fe_2O_3 By The Following Equation. Mar 2th,

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Simulation The Skater Had More Potential Energy At

The Bottom And Had More Kinetic Energy At The Top.

The Relationship Between Kinetic And Potential Energy

Is That When One Goes Up, The Other Comes Down.

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