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Answers: Moles And Stoichiometry Practice ProblemsAnswers: Moles And Stoichiometry Practice Problems 1) How Many Moles Of Sodium Atoms Correspond To 1.56×10^{21} Atoms Of Sodium? 1.56×10^{21} Atoms Na $\times \frac{1 \text{ Mol Na}}{23.00 \text{ g Na}} = 6.78 \times 10^4 \text{ mol Na}$ $\times 23.00 \text{ g/mol Na} = 1.56 \times 10^6 \text{ g Na}$ 2) Determine The Mass In Grams Of Each Of The Following: A. 1.35 Mol Of Fe $1.35 \text{ Mol Fe} \times 55.845 \text{ g/mol Fe} = 75.4 \text{ g Fe}$ 1 Mol Fe B. 24.5 Mol O $24.5 \text{ Mol O} \times 16.00 \text{ g/mol O} = 392 \text{ g O}$ Jan 15th, 2024Stoichiometry Practice Problems Pdf AnswersStoichiometry Practice Problems Pdf Answers Balance The Following Chemical Reactions: A. $2 \text{ CO} + \text{O}_2 \rightarrow 2 \text{ CO}_2$ B. $2 \text{ KNO}_3 \rightarrow 2 \text{ KNO}_2 + \text{O}_2$ C. $2 \text{ O}_3 \rightarrow 3 \text{ O}_2$ D. $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \dots$ Apr 13th, 2024Stoichiometry Practice Problems Worksheet 1 AnswersNov 15, 2021 · Stoichiometry-practice-problems-worksheet-1-answers 2/2 Downloaded From Edu-dev.fuller.edu On November 15, 2021 By Guest Earth's Crust Can Often Be Iden Jan 6th, 2024.

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