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PERFORM THE FOLLOWING PERCENT YIELD CALCULATIONS: 18. If 12.5 G Of Copper React With Excess Chlorine Gas, Then 25.4 G Of Copper(II) Chloride Are Produced. Find The Theoretical And Percent Yields. $\text{Cu} + \text{Cl}_2 \rightarrow \text{CuCl}_2$ 19. If 6.57 G Of Iron React With An Excess Of Hydrochloric Acid Jan 15th, 2024

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